

## VERSITY, SRI LANKA FIRST YEAR FIRST SEMESTER EXAMINATION IN SCIENCE-2012/2013 (Feb' 2014) CH 151 QUANTITATIVE AND QUALITATIVE INORGANIC ANALYSIS

Group 2

You are provided with a mixture A containing two inorganic cations. Analysis the mixture A qualitatively and record your observations, inferences and conclusion. Carryout one confirmatory test for each identified cation.

Hint: Assume the cations are present in Group II and Group V only.

- A mixture B contains two inorganic anions. Perform the following tests and record your observations, inferences and conclusion. Carryout one confirmatory test for each identified anion.
  - a) Add dil. H<sub>2</sub>SO<sub>4</sub>, warm and test for evolved gas
  - b) Prepare Na<sub>2</sub>CO<sub>3</sub> extract and use the extract to the following experiments.
    - i. Add dil. HNO3 and AgNO3
    - ii. Add dil. HNO3 and BaCl2
    - Add dil. HCl and H2S iii.

**Time: Three Hours** 

- Boil with few drops of con. HCl and pass H<sub>2</sub>S iv.
- Add few drops of NaOH to the extract and then test with fresh dil. V. Sodium nitroprusside.
- Boil with con. HNO<sub>3</sub> and ammonium molybdate vi.
- Acidify the Na<sub>2</sub>CO<sub>3</sub> extract with dil. H<sub>2</sub>SO<sub>4</sub> and add freshly prepared vii. FeSO<sub>4</sub> and few drops of con. H<sub>2</sub>SO<sub>4</sub>.