



Eastern University, Sri Lanka

Third Year Second Semester Examination in Science

2010/2011 (September/ October 2015)

CH 306 Surface Chemistry and Molecular spectroscopy (Proper)

Answer all questions

Time: 01 hour

Velocity of light (c) =2.99x10⁸ m s⁻¹ Gas constant (R) = 8.314 J K⁻¹ mol⁻¹

Plank's constant (h) = $6.626 \times 10^{-34} \text{J s}$

- 1.
- (a) i) Write the BET adsorption isotherm and identify the terms in this equation.
 - ii) Show that at low pressures and for large values of 'c 'this equation reduces to Langmuir equation and explain this behavior.

(30 marks)

(b) For the adsorption of N₂ on 1.00 g of ZnO the volumes (converted to 0⁰ C and 1 atm) of N₂ adsorbed were 1.06 and 2.08 cm³ g⁻¹ when the equilibrium pressures were 183 and 533 mmHg respectively. Assuming Langmuir behavior,

ii) If N₂ molecule covers $1.2 \times 10^{-18} \text{ m}^2$, calculate the surface area of a 1 g sample of ZnO.

(45 marks)

(c) The contact angle for water on clean glass is close to zero. Calculate the surface tension of water at 20° C. At that temperature water climbs to a height of 4.96 cm in a clean glass capillary tube with internal radius of 0.300 mm. The density of water at 20° C is 998.2 kg m⁻³.

(25 marks)

i) Calculate the equilibrium constant (or adsorption constant)

- (a) Consider the $v = 2 \rightarrow 3$ rotation –vibrational band of the ${}^{11}B^{19}F$ molecule within the harmonic oscillator, rigid rotator approximation. The vibrational frequency and rotational constant of this isotopomer (${}^{10}B^{19}F$) are equal to 1402.13 cm⁻¹ and 1.50724 cm⁻¹, respectively.
 - i) Write the energy expression for rotation -vibrational spectra of harmonic oscillator
 - ii) Determine the frequency or spectral position (in cm⁻¹) of the P- branch transition with $J = 4 \rightarrow 3$
 - iii) Calculate the vibrational frequency (cm⁻¹) for ${}^{10}B^{19}F$
 - iv) Predict the equilibrium bond length of BF

(75 Marks)

(b) Briefly explain the quantum theory of Raman Effect in terms of originating stokes, anti-stokes lines in the spectrum.

(25 Marks)