## EASTERN UNIVERSITY, SRILANKA SECOND EXAMINATION IN SCIENCE (FIRST SEMESTER) 2001/2002 CH203 SPECTROSCOPIC METHODS

Time: 01 Hour

Answer all questions.

1.(a) Using woodward-fiesher-scott rule calculate the  $\lambda_{max}$  value for each of the following compounds.

 $\int$  (b) Sketch the Proton NMR spectrum including the multiplet expected for each of the following compounds with TMS as a standard. Predict the approximate chemical shift in each of the spectra.

- (c) The InfraRed spectrum of a compound <u>A</u> (C<sub>10</sub>H<sub>12</sub>O<sub>2</sub>) showed strong absorptions at 1720 cm<sup>-1</sup>;1100 cm<sup>-1</sup>·1320 cm<sup>-1</sup> 1600 cm<sup>-1</sup>,690& 750 cm<sup>-1</sup>. H NMR spectrum of the compound A had signals at δ 7.9(m,2H), δ 7.5(m;3H) ,δ 5.2 (septet,1H) , δ1.4(d,6H). Interpret the spectral data and deduce the structure of the compound <u>A</u>.
  - (d) Explain the term 'Mc.Lafferty rearrangement' with an example.

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