

Second Year Second Semester Examination in Science 2008/2009(October 2010)- Repeat CH 203 Spectroscopic Methods

Time Allowed: ONE HOUR

Answer all the questions

 (a) How would you differentiate the following isomeric pairs by UV-Visible Spectroscopy.

(b)Explain the following observation

30 marks

 $\lambda_{\text{max}} = 217 \text{nm}$ 

 $\lambda_{\text{max}} = 263 \text{nm}$ 

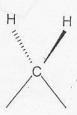
20 marks

(c) (i) IR spectrum of benzaldehyde shows absorption band at V /cm<sup>-1</sup> 3100, 2840, 2730, 1700, 1600,750 and 690. Interpret the spectral data to the characteristic group vibrations of the molecule.

Benzaldehyde

25 marks

(ii) Draw the pictorial diagram to show the various vibrational modes for a CH<sub>2</sub> group



2. (a) Give the number of 1H-NMR signals that could be observed for the following compounds

(b) Sketch the <sup>1</sup>H-NMR spectrum of the compound shown below.

40 mar

(c) Explain the following observation (ie. Give the fragmentation mechanism)

$$\begin{matrix} \mathsf{O} \\ || \\ \mathsf{CH_3} \, \mathsf{CH_2} \, \mathsf{CH_2} \mathsf{C} \longrightarrow \mathsf{CH_3} \end{matrix}$$

2-pentanone m/z 86, 71, 58, 43 and 29

30 mar