



# EASERN UNIVERSITY, SRI LANKA

THIRD EXAMINATION IN SCIENCE – PROPER

SECOND SEMESTER 2004/2005 (OCTOBER 2006)

## CH 304 QUANTUM CHEMISTRY & METALLURGY AND INDUSTRIAL CHEMISTRY

Time allowed: **ONE Hour**

---

*Candidate must NOT start writing their answers until told to do so*

You may find the following data useful

Avogadro constant ( $N_A$ ):  $6.023 \times 10^{23} \text{ mol}^{-1}$

Electron charge (e):  $1.602 \times 10^{-19} \text{ C}$

Faraday constant (F):  $9.648 \times 10^4 \text{ Cmol}^{-1}$

Gas constant (R):  $8.314 \text{ JK}^{-1}\text{mol}^{-1}$

Planck's constant (h):  $6.626 \times 10^{-34} \text{ Js}$

Rest mass of electron ( $m_e$ ):  $9.1 \times 10^{-31} \text{ kg}$

Velocity of light (c):  $3 \times 10^8 \text{ ms}^{-1}$

The use of a non-programmable calculator is permitted

1. (a) (i) Starting from the one-dimensional wave equation  $\frac{\partial^2 u}{\partial x^2} = \frac{1}{v^2} \frac{\partial^2 u}{\partial t^2}$  for the standing wave system and substituting  $u(x,t) = \psi(x) \cos(\omega t)$ ,  $\omega = 2\pi\nu$  and  $v\lambda = v$ , show that

$$\frac{\partial^2 \psi(x)}{\partial x^2} + \frac{4\pi^2}{\lambda^2} \psi(x) = 0 \quad (20 \text{ marks})$$

- (ii) Using the derivation obtained in (i), derive the following time independent Schrodinger equation for a particle moving in a one - dimensional box with the help of de Broglie relation and the relation of momentum  $p = \sqrt{2m(E - V)}$

$$\frac{d^2 \psi(x)}{dx^2} + \frac{8\pi^2 m}{h^2} (E - V) \psi(x) = 0 \quad (10 \text{ marks})$$

- (b) (i) Calculate the probability that a particle restricted to move in a one - dimensional box of length 'a' is found to be the distance between '0' and 'a/4'. The wave function ( $\psi$ ) of the particle given by  $\sqrt{\frac{2}{a}} \sin\left(\frac{\pi x}{a}\right)$

(20 marks)

- (ii) What is the probability of the particle beyond the distance 'a/4'?

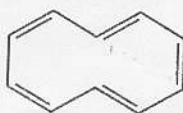
(10 marks)

- (c) (i) Write the general expression for the energy levels of a particle of mass 'm' moving in a two - dimensional square box of length 'a' and identify all terms in it.

(10 marks)

- (ii) The  $\pi$  electrons of a 10 - Annulene may be modeled as electrons in a two-dimensional square box. If the edge length of the box is 1000 pm and the  $\pi$  - system contains 10 electrons, what is the energy of the HOMO - LUMO transition? ( $1 \text{ pm} = 10^{-12} \text{ m}$ )

(30 marks)



10 - Annulene

2. (i) Metals can be extracted from their ores through several processes. Give a short account on refining of metals (physical and chemical methods). (35 marks)
- (ii) Briefly explain the dry process of manufacture of Portland cement. (35 marks)
- (iii) Briefly explain the important steps in Glass forming process. (30 marks)

End