

EASTERN UNIVERSITY, SRI LANKA
THIRD EXAMINATION IN SCIENCE - 2005/2006
SECOND SEMESTER (PROPER/REPEAT)
(MARCH/APRIL 2008)



PH 305 FUNDAMENTALS OF STATISTICAL PHYSICS

Time: 01 hour.

Answer ALL Questions

1. Derive the relation between the thermal average energy and the single particle partition function for a system of N non-interacting, distinguishable particles.

A system at temperature T consists of N non-interacting distinguishable identical particles each of which can be in either $+\varepsilon$ or $-\varepsilon$ energy states, but the particles do not have any translational kinetic energy.

- i. Write down the partition function for a single particle.
- ii. Find the thermal average energy of the system.
- iii. Given that the total energy of the system is U , show that the absolute temperature is:

$$\frac{1}{T} = \frac{k}{2\varepsilon} \ln \left[\frac{N - \left(\frac{U}{\varepsilon}\right)}{N + \left(\frac{U}{\varepsilon}\right)} \right].$$

- iv. Obtain an expression for heat capacity C_V of the system.
2. Derive the expressions for thermodynamic probability and the distribution function in the case of Bose-Einstein statistics. Hence explain how the classical result can be obtained from Bose-Einstein statistics.

Show that for an ideal gas, the partition function Z can be written as

$$Z = 2\pi V \left(\frac{2mkT}{h^3} \right)^{\frac{3}{2}} \frac{\sqrt{\pi}}{2}$$

You may use the following identity

$$\int_0^{\infty} x^{\frac{1}{2}} e^{-x} dx = \frac{\sqrt{\pi}}{2}$$