EASTERN UNIVERSITY, SRI LANKA THIRD EXAMINATION IN SCIENCE (FIRST SEMESTER)- 2002/2003 CH 301 CHEMISTRY OF NATRUAL PRODUCTS

Time: 01 Hour Answer all questions

(1). Answer all three parts (a), (b), and (c)

(a). Draw the Fischer projection formulae of Ribose, Arabinose sugars having the D-

A D (+)- aldopentose $\underline{\mathbf{A}}$ (C₅H₁₀O₅) on treatment with sodium borohydride (NaBH₄) gives an optically active compound $\underline{\mathbf{B}}(C_5H_{12}O_5)$. Treatment of $\underline{\mathbf{A}}$ with bromine water gives $\underline{\mathbf{C}}$ (C₅H₁₀O₆), which when heated with concentrated ammonia gives $\underline{\mathbf{D}}$ $(C_5H_{11}NO_5)$. Treatment of $\underline{\mathbf{D}}$ with bromine in sodium hydroxide gives $\underline{\mathbf{E}}$ $(C_4H_8O_4)$, which reacts with sodium borohydride giving an optically inactive compound F $(C_4H_{10}O_4).$

Deduce the structures of the compounds A, B, C, D, E and F. By means of equations show how the compound $\underline{\mathbf{A}}$ could be converted to D (+)-fructose

b). (i). Name all the sugars that will yield the same osazone as D-glucose.

(ii). Show how D- glucose forms an osazone upon reaction with an excess of Phenyl hydrazine (PhNHNH₂) (your answer should include mechanism of the reaction)

e). By means of equations show how two of the following conversions maybe effected.

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2. Answer all three parts (a), (b), and (c)

- (a). Give the isomeric natures of the Citral and discuss a method to synthesis one its
- (b). (i). Show how Phenyl isothiocyanide (Ph-NCS) could be used to determine the Nterminal residue in a peptide.
 - (ii). Indicate by means of equations, how the following transformation maybe

PhCH₂SH
$$\longrightarrow$$
 S -CH₂-CH-NH₂
S -CH₂-CH-NH₂
COOH

- (c). (i). Why the esterification reactions occurs readily in menthol than in neomenthol.
 - (ii). Write down the structure of cholesterol and explain two colour reactions specific for cholesterol
 - (iii). By means of equations, show how the following transformations could be effected. Give essential experimental conditions.

(i).
$$CO_2Et$$
 CO_2H CO_2H CO_2H CO_2Et CO_2ET